

## **Year 10 Chemistry**

Students who opt to follow the triple science route continue to build on the knowledge and understanding gained from Year 9. Topics studied in Year 9 align with the GCSE Chemistry AQA specification so the transition from Year 9 to 10 is smooth. Through Year 10 students study topics that build in complexity and utilise learning from earlier topics. C9 develops ideas about our atmosphere, its composition and how it has evolved over time linking to atmospheric pollutants. C10 develops students' understanding of the Chemistry behind the earth's natural resources that we utilise every day. In C1 we then build on students' prior knowledge from Key Stage 3 on both the structure of the atom and the periodic table. C2 finishes the year off by building on their knowledge of the atom to look at the structure and bonding within a variety of different substances.

Investigative skills are developed by a number of required practical activities. There are also other practical activities that enriches the learning, as well as practical demonstrations that do the same. Students are assessed at the start and end of the year with tests that check learning on topics in Year 9 and Year 10. This continues to build in demand as students progress through the GCSE course as more and more content is covered in the test. Additionally there is a PPE in April which has a similar arrangement to the other assessments but takes place in a more formal setting. Students are taught by one teacher in 5 lessons per fortnight

Methods of deepening and securing knowledge:				
Retrieval practice	Almost all lessons have retrieval practice in them. This is usually as a starter activity.			
Interleaving	Retrieval practice includes interleaved questions from previous topics, making connections between topics where possible.			
	Many ideas from Key Stage 3 are revisited during Year 10 lessons.			

	Autumn term 1	Autumn term 2	Spring term 1	Spring term 2	Summer term 1	Summer term 2	
Topic(s)	C9 Chemistry of the Atmosphere		C1 Atomic structure and the periodic		C2 Bonding, structure and the		
	- Proportions of diffe	- Proportions of different gases in the		table		properties of matter	
	Earth's atmosphere	Earth's atmosphere		- Atoms elements and compounds		- Chemical bonds	
	- The earth's early atmosphere		- Mixtures		- Ionic bonding		
	- How oxygen increa	sed	- Development of the model of the atom		- Ionic compounds		
	<ul> <li>How carbon dioxide decreased</li> <li>Greenhouse gases</li> <li>Human activities which contribute to an increase in greenhouse gases in the atmosphere</li> </ul>		- Relative electrical charges of subatomic		- Covalent bonding		
			particles		- Metallic bonding		
			<ul><li>- Size and mass of atoms</li><li>- Relative atomic mass</li></ul>		- The three states of matter		
					- State symbols		
			- Electronic structure		- Properties of ionic compounds		
	- Global climate char	- Global climate change		- The periodic table		- Properties of small molecules	
	- The carbon footprint and its reduction - Development of the periodic table - Atmospheric pollutants from fuels - Metals and non-metals		e periodic table	- Polymers			
			- Metals and non-me	on-metals - Giant covalent structures		uctures	

	Proportios and offects of atmospheric	Group 0	Droportios of motals and alloys	
	- Properties and effects of atmospheric	- Group 0	- Properties of metals and alloys	
	pollutants	- Group 1	- Metals as conductors	
		- Group 7	- Diamond	
	C10 Using the earth's resources and	- Comparison of transition elements with	- Graphite	
	obtaining potable water	Group 1 elements	- Graphene and fullerenes	
	- Using the earth's resources and	- Typical transition element properties	- Sizes of particles and their properties	
	sustainable development	,,,	(nanoparticles)	
	- Potable water		- Uses of nanoparticles	
	- Waste water treatment		Oses of Harloparticles	
	- Alternative methods of extracting metals			
	- Life cycle assessment			
	- Ways of reducing the use of resources			
	- Corrosion and its prevention			
	- Alloys and useful materials			
	- Ceramics, polymers and composites			
	- The Haber process			
	- Production and uses of NPK fertilisers			
	Troduction and ases of the Richmsers			
Assessment	Aiming High 1 test	PPE: a whole exam	Aiming High 3 test	
	including topics	paper covering C6-	covering all Year	
	studied up to this	10	10 topics	
	point in Year 10 and			
	the Year 9 topics			
	the real 5 topics			

## Independent Learning:

Independent learning is set every week in line with the school policy. We set online work on Educake which reviews previous learning in a quiz-like format. Additionally we utilise 'knowledge organiser' booklets to set students tasks to produce revision material from allocated pages. We also set 'language for learning' tasks every half term which requires students to complete a quiz on tier 2 and 3 words encountered in the science course.